

Application 10/569,554

Filing date: 08/19/2004

Inventors: FARRUSSENG David, MIRODATOS Claude, REBEILLEAU Michael, VAN VEEN André, RUSHWORTH Simon and ROUSSET Jean Luc

**DECLARATION under 37 C.F.R. § 1.132**

The undersigned, David FARRUSSENG of 136H, rue Guillaume de Varey, 69380 BELMONT D'AZERGUES, France, herewith declares as follows :

1. I am an expert in the field of material chemistry. My resume is attached.
2. I am a co-inventor of the U.S. patent application serial number 10/569,554. I am aware of the contents of the pending claims. I have reviewed the Office Actions mailed on 10/23/2008, 01/14/2009 and 07/30/2009, the advisory action dated 12/17/2009 and the prior art cited therein.
3. The invention of the application includes an oxygen conducting membrane, comprising a mixed conducting dense membrane of multmetal oxide, one surface of which is covered with dispersed particles based on magnesium oxide or noble metals,  
wherein the mixed conducting dense membrane is a layer comprising a multmetal oxide having the formula:  

$$\text{Ba}_x \text{Sr}_{1-x} \text{Co}_{1-y} \text{Fe}_y \text{O}_{3-z}$$

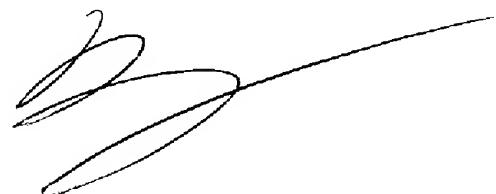
where  $0 \leq x \leq 1$ ;  $0 \leq y \leq 1$  and  $z$  is a number which renders the charge of the compound neutral and which defines the oxygen deficiency, and  
wherein the particles based on magnesium oxide or noble metals represent from 0.01 to 0.1% by weight of the dense membrane.
4. The claimed oxygen conducting membrane differs from the one of Chen et al. by the nature of the multmetal oxide and by the weight ratio of metal or metal oxide compared to the one of the dense membrane of Chen et al. (US 6,187,157).
5. The claimed oxygen conducting membrane is not obvious in view of the teaching of Chen et al. or in view of the teaching of Chen et al. combined with the teaching of Hazbun (US 4,791,079).
6. The claimed membranes advantageously exhibit better oxygen fluxes than membranes of the prior art. The claimed membranes usually allow an increase of the permeation flow of the oxygen, generally of between 5 and 20%, and advantageously in the order of 30% (p. 9, 3<sup>rd</sup> par.of the present application).
7. I declare that all statements made in the enclosed article by Olivier et al. (Catalysis today, 2009, vol. 142, n° 1-2 pp. 34-41) of my knowledge are true. Table 1 describes a 1 mm thick Pt/MgO coated  $\text{Ba}_{0.5}\text{Sr}_{0.5}\text{Co}_{0.8}\text{Fe}_{0.2}\text{O}_{3-z}$  membrane (table 1, p. 36). The exemplified membrane contains 7.2 mg of Pt/MgO (table 1) for approximately 0.5 g

perovskite (p. 35, col. 1 "membrane materials"), i.e. about 0.014%wt of Pt/MgO. This membrane responds to the definition of the claimed membrane. As reported in table 1, the oxygen flux at 900°C of such a membrane is 2,84 mL/cm<sup>2</sup> min.

8. This oxygen flux is 40% higher than the best oxygen flux reported in Chen *et al.*, which describes that the oxygen flux at 900°C of the membrane of example 1 (Ag coated La<sub>0,05</sub> Sr<sub>0,95</sub> Co O<sub>3-x</sub>) is 2 mL/cm<sup>2</sup> min (from figure 4, for a 1 mm thickness) and the one of example 3 (50Pd/50Ag coated Ce<sub>0,8</sub> Gd<sub>0,2</sub> O<sub>2-x</sub>) is 0,06 mL/cm<sup>2</sup> min.
9. This demonstrates that the claimed oxygen conducting membrane exhibits better oxygen fluxes than the membranes of Chen *et al.*
10. Said improvement of the oxygen fluxes by using a membrane as claimed could not have been foreseen from the teaching of Chen *et al.* taken alone or in combination with that of Hazbun *et al.*
11. The undersigned declares further that all statements made herein of his knowledge are true and that all statements made on information and belief are believed to be true. Further that these statements were made with the knowledge that wilfully false statements and the like so made are punishable by fine or imprisonment or both under Section 1001 of Title 18 of the United States Code, and that such wilfully false statements may jeopardize the validity of the application and of any patent issued thereon.

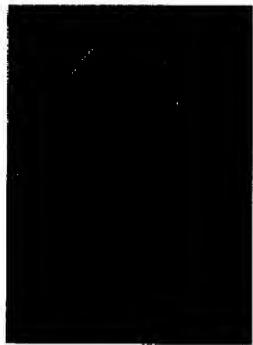
Signed this 18 day of January 2010.

David FARRUSSENG



## David Farrusseng

Born in 1972 at Montpellier, France  
Married, three children



**1994** Master I at Montpellier University

**1994** Stay at Heidelberg University, Prof W. Siebert, *Titanocene synthesis*

**1995** Master II, Montpellier University, Prof. R. Corriu, *Alkoxy silane synthesis*

**1996-1999** PhD thesis, Institut Européen des Membranes (Univ. Montpellier) –

Supervisors: Christian Guizard & Anne Julbe -*Catalytic membranes for the partial oxidation of alkanes.*

**2000** Post-Doc at Max Planck Institute für Kohlenforschung, Prof. Ferdi Schueth, Synthesis of micro & mesoporous silicon nitride for basic catalysis

**2001** Permanent staff at Institut de Recherches Catalyse (IRCELYON), Lyon, France

**2010** Group leader at IRCELYON/CNRS

### Selected papers

*A comparative study of  $La_{0.8}Sr_{0.2}MnO_3$  and  $La_{0.8}Sr_{0.2}Sc_{0.1}Mn_{0.9}O_3$  as cathode materials of single-chamber SOFCs operating on a methane-air mixture*  
Journal of Power Sources, 2009, 191(2): p. 225-232.

*Metal-Organic Frameworks: State of the Art and Opportunities for Catalysis*  
Ang. Chem. Int. Ed., 2009, 48(41): p. 7502-7513.

*A new symmetric solid-oxide fuel cell with  $La_{0.8}Sr_{0.2}Sc_{0.2}Mn_{0.8}O_{3-d}$  perovskite oxide as both the anode and cathode*  
Acta Materialia, 2008, 57(4): p. 1165-1175.

*High Throughput Experimentation in heterogeneous Catalysis*  
Surface Science Reports, 2008, 63(11): p. 487-513.

*Limitations and potentials of oxygen transport dense and porous ceramic membranes for oxidation reactions*  
Catalysis Today, 2005, 104(2-4): p. 102-113.

*Pore-size engineering of silicon imido nitride for catalytic applications*  
Angewandte Chemie-International Edition, 2001., 40(22): p. 4204-4207.

*The first redox switchable ceramic membrane*  
Journal of the American Chemical Society, 2000, 122(50): p. 12592-12593.

### Scientific outcome

9 plenary invited lectures in international conferences, 25 oral presentation in international conferences, 60 publications in international journals, 8 patents (1 exploited), 2 software, 4 book chapters, board member of CCHTS, more than 110 conference proceedings, and more than 600 citations.

### Awards and Honours

**1995** Grant assigned to for best students

**1999** Award for outstanding PhD work, Montpellier University

**2007** DivCat Annual Award of the Catalyst Division of French Chemical Society